I claim:

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1 1. A method of removing sulphide ion from a fluid having a pH in excess of about 9 comprising:

adding a ferrous gluconate chelating agent to said fluid in sufficient quantity to form iron sulphide with the sulphide ion; and

mixing the chelate with the fluid and forming iron sulphide.

- 1 2. The method of claim 1 further comprising maintaining the ferrous gluconate at a level to maintain the sulphide concentration below a certain desired level.
- The method of claim 1 wherein the quantity of gluconate added to said fluid exceeds the quantity needed to chelate all of the iron in said fluid.
 - 4. A method of reducing the hydrogen sulphide concentration in a drilling fluid comprising: adding a ferrous gluconate compound to said fluid; and allowing said ferrous gluconate to react with said hydrogen sulphide such that sulphide is precipitated.
 - 5. The method of claim 4 wherein said sulphide is precipitated as iron sulphide.
 - 6. The method of claim 4 wherein said drilling fluid has a pH greater than 9.0.
- The method of claim 4 wherein said drilling fluid has a pH ranging from about 11 to about 12.
- 1 8. A drilling fluid additive comprising an iron (II) based hydrogen sulphide scavenger chelated
- with a gluconate chelating agent which provides a stable complex with said iron at pH greater than
- 3 about 9.

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1 9. The drilling fluid additive of claim 8 which provides a stable complex with said iron at a

- pH of at least about 11.5.
- 1 10. The drilling fluid additive of claim 8 which provides said stability at subterranean formation
- 2 temperatures.
- 1 11. The drilling fluid additive of claim 8 which provides said stability at temperatures ranging from ambient temperature to over 300 degrees Fahrenheit.
- 1 12. The drilling fluid additive of claim 8 which provides improved resilience to the rheological properties of said fluid.
 - 13. In combination with a drilling mud comprising crosslinkable polymers delivered to a well during drilling operations, an additive for decreasing hydrogen sulphide concentration in the mud, the additive comprising an effective amount of a ferrous chelating agent mixed into the mud to reduce hydrogen sulphide concentration in the mud circulating in the well, wherein such additive fails to significantly cause crosslinking of said polymers.
 - 14. The additive of claim 13 wherein the iron in said ferrous chelating agent does not significantly ionize to a trivalent state in said mud.
 - 15. The additive of claim 13 wherein said additive enhances the mud's ability to withstand well temperatures under shear.
- 1 16. A polymer based drilling fluid comprising ferrous gluconate.